

## Uploaded to VFC Website ~ October 2012 ~

This Document has been provided to you courtesy of Veterans-For-Change!

Feel free to pass to any veteran who might be able to use this information!

For thousands more files like this and hundreds of links to useful information, and hundreds of "Frequently Asked Questions, please go to:

## Veterans-For-Change

Veterans-For-Change is a 501(c)(3) Non-Profit Corporation Tax ID #27-3820181

If Veteran's don't help Veteran's, who will?

We appreciate all donations to continue to provide information and services to Veterans and their families.

https://www.paypal.com/cgi-bin/webscr?cmd= s-xclick&hosted button id=WGT2M5UTB9A78

Note

VFC is not liable for source information in this document, it is merely provided as a courtesy to our members.

item ID Humber	01477
Au <b>ther</b>	
Corporate Author	
Report/Article Title	Manuscript: Exposure calculations
Journal/Book Title	
Year	0000
Month/Bay	
Cefer	[
Kumber éf Inzgés	85

Includes notes and calculations on various exposure scenarios. Items were filed in one folder labeled "Exposure Calculations"

**Descripton Notes** 

ACGIH 8-W TWA TLV for 2,4-D or 2,4,5-T 10 mg/m3 If 2,4,5-T now has TCDD @ O.Img/kg then the implied std 15 10 mg 2, 4, 5-T x 0.1 mg TCDD x Kg 2, 4, 5-T x 106 mg  $= 1 \times 10^{-6} \, \text{mg TCDD} = \frac{1 \, \text{mg TCDD}}{1 \, \text{mg TCDD}}$ 

 $\frac{1 \text{ Mg TCDD}}{\text{M}^{3}} \times \frac{1.8 \text{ m}^{3}}{\text{Nu}} \times \frac{8 \text{ lu}}{\text{day}} = \frac{14.4 \text{ mg}}{\text{day}}$   $\frac{14.4 \text{ mg}}{\text{day}} \times \frac{\text{man}}{70 \text{ kg}} = 0.21 \frac{\text{mg}}{\text{kg}}$ 

2

Respiration rate for Average Man at work

30 resp/min }
 500 ml / resp. }

.. 30 resp x 500 ml x min x 500 ml x 106 ml x day

= 7.2 m3/day

Assume average man 15 72 Kg

No TCDD was detected in our samples (See Ch2)

Taking worst case of minimum detectable concentration (MDC) of = 36 mg/m³ and assuming levels of TCDD were just equal to this moc

36 mg x 7.2 m³ = 259,2 mg TCDD m³ 8.h. Day

Assume men 13 72 Kg

259.2 mg TCDD X Man = 3.6 mg TCDD 8-m Day X 72 Kg Man Kg. Day

4/

Use worst case of 2.4.5-T noted at NCBC along with worst case of TCDD in 2,4,5-T as 94 mg TCDD/Kg 2,4,5-T, also use mean TCDD in 2,4.5-T as Amg/Kg

Worst Case: 79.62 ug 2.4.5-T/m3
@ NCBC

29.62 Ug 2,4,5.T x 94 mg TCDD x Kg 109 Ug

= 7.48 ×10-6 mg TCDD

= 7.48 <u>mg TCD A</u> m3.

Assuming unprotected worker respiring 30 x/min @ 500 ml/resp = 7.2 m3/day

.. Total inhaled =

7.48 mg TCDD x 7.2m 3 man = 0.748 mg m3 day x 72kg = 0.748 mg

2 This 13 - 1.0 mg Kg. Day

5/

Using: 4 mg TCDD/Kg 2,4,5-T high Mean 32.39 Ug 2,4,5-T m3

Sto Ber 21.02

95% Mean + 25D - 74. 549

HO - 32 Mg T

Inhalation exposures Direct Exposure to spray (Using RPAR approach)

1) 0.067 mg/m<sup>3</sup> x 12# 2,4.5-T = 1.748 mg/m<sup>3</sup> 0,46#/A

2) Resperable fraction "worst care" - is "/6 (ree/RPAR) - 604 size -

(1.748 mg/m3)(1/6) = 0,291 mg/m3

at a verpuation rate of 1.8 m3/hr and a duration of exposure of theo. the total absorbed (assuming 100% absorption of all inhaled) mound be be  $0.291 \, \text{mg} \times \frac{18 \, \text{m}^3}{\text{m}^3} \times \frac{2 \, \text{hr}}{\text{day}} = 1.0476 \, \text{mg/day}$ 

1.0476 mg/day

Assume & cases

1) TCDD mean conc Putple max 94 mg/kg

2) " " " Hean 32 mg/kg

3) " " HO Max 14 mg/kg

4 " " " Mean 2 mg/kg

then the TCDD exposures would be, 9 4 mg/Kg 32mg/49 14, mg/Kg Zing/Kg 1.0476 mg 2.4.5-T x 94 mg 7.4.5-T = = 33,52 H X 32 2) = 14.67" × 14 4.095" × 4 Assuming a 70 kg man the inhalad dose would be 2.4.5-T 1.0476 Mg X TOK9 = 1543-45-T Day K9

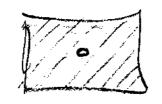
.

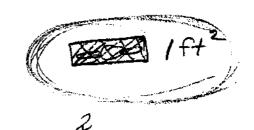
•		0,012 0,0025 0,0007		
		0,012	0.0025	0.0007
		100% penet.	79% penet	690 pend
6 TCDD		Dermal	+ Inhal	•
Conna n	TC b D (mg	ilka:Dan)		
Care	1.407	13.41	3.91	2.11
Z	0,479	12.48	2.98	1.18
3	0,209	12.21	2.71	0.91
4	0.030	12.03	2,53	0.73

There calculations do not take into account the degredation that would occur in TCDD during application.

% dermal dose to Total

Care	_	lono ha	100% P 89	29%	6%P
1	Pounle	94	89	64	33
7	n	32	96	84	59
_	HO		98	92	77
_	Ŋ		99	99	96 .





3.556 mg M 0.46 # M 0.76 gol/Acre

(3.556 mg M) (0.76 gab/A) x 4\*T RPAR

2.35 mg T

3.55b mg M (0.76 spl H.0/A) x 4# T RUN

0.46 # M gal HO

(23.5 mg T)

0.46 # M x 4549 x 1000mg # 1000mg 48,560 ft<sup>2</sup>/A

4.83 mgM

i.e.

A

Caplan et al. (167 in RPAR) working with aerially applied mulathion in oil sprays explied at 0.46 pounds per 0.76 gallone water/acre, determined a dermal exposure to serious ducetly hereally the spray plane for have shin ( head, nech, shouldow forground, hands and thighs) of 3.556 mg/day. Using there data, an equivalent dermal exposure for 2,4,5-T and TCDD, aerually applied at 12 pounds aced equivalent 3,4,5-T per 3 gallom of Kerbucede Orange / acre can be determined

3

The following relations will give the baily worst-case daily dermal exposure for 2,4,5-T and TCDD:

- D 3,556 mg Mola x 12 # = 92.8 mg 2.4,57
- 2) with a TCDD worst case
  of 94 mg TCDD/Kg 2,4,5-T, the
  TCDD dermal exposure rate per
  day would be

92.8 mg 2.4.5-T x 94 mg TCDD x Kg 2.4.5-T 10°mg

= 8,72 ×10-3 mg = 8.72 ug TCDD

3) Assuming 10% of materal 13 absorbed

2,4.5-T dermal exposure = 9.28 mg TCDD dermal exposure = 0.87 ug

- 4) For 70 kg man (w/o could could
  - (a) a.4.5-T dermal dose rate = 0.133 mg/KgTCDD n n = 0.012 Ug/Kg

See Ch 1 5) Assuming canopy penetration rate is 21%, the dermal dose rates are

(b) 2,4,5-T 0.0279 mg/Kg
TCDD 0.0025 Ug/Kg

of essuming canopy penetration rate is 600, the dermal dose rates are

© 2,4,5-T 0,0080m9/K9 7200 0,0007 ug/K9 Inhalation Exposure aerial

 $\frac{0.067 \, \text{mg/m}^3}{.46 \, \text{ft}} \times 12 = 1.748 \, \text{mg}$   $1.748 \, \times \frac{1}{6} = 0.291 \, \text{mg/m}^3$ 

Resp Rate = 7.2 m3/day = 0.9 m3/hr

291 mg x 0.9m3 x 8hi = 2.09 mg

an stu

2,4,5-T = 2.09 mg - 0.029 mg Kg. Day

aruning 94 mg/Kg 2,4.5.T (ug/g)=(mg/mg)

0.029 mg 2.4.5-T × 94 mg TCDD = 2.814 mg Kg-Day mg 2.4.5-T = Kg-Day But Canopy penetration ~ 21% 2,4.5-T exporure 6.09 Mg /Kg Day (.029 mg)(.21) TCDD exposure 2.814mg (121) 0.59 Mg/Kg. Day Cumulatine exposure = TCDA Skin + Inhal 2.5 Mg 27.9 Hg /Kg 0.59 mg/Kg 6.09 Ug/Kg 3.09M9/Kg 33.99Ug/Kg 20 mg/Kg PEL

30 mg/Kg

HP

Worst care

Hest. Purple & TCOD = 47mg/kg
10% skin absorption
100% canopy penetration
70Kg man
16 respirable fraction 60H size
1.8 m3/h respiration rate, 100% abe.
2-h exposure

Dermal Exposure

From RPAR

3.556 mg Malathuril x 12# 2,4,5.T

= 93 mg 2,4,5-T

- 10% skin absorption = 9.3 mg 2.4.5-T

- TCDD 9.3 mg 2.4.5-T x 94 mg TCDD x Kg 2.4.5-T x 106mg

= 0,87 Mg TCDD

HP

## Inhalation + Dermal Exposure

2,4,5-T

Inhal Rermal

1.05 mg 9,3 mg 10.35 mg

For 70 kg man

0.148 mg 1

TCDD

Inhal Dermal

0.098 Ug 0.87 Ug 0.968

For 70 Kg

0.968 Ug TCOD = 0.014 Ug TCOD Kg

HP

Inhabation Exposure From RPAR

 $\frac{0.067 \, \text{mg/m}^3}{0.46 \, \text{#}} \times 12 = 1.75 \, \text{mg/m}^3$ 

Respirable fraction 'worst case"  $(1.75 \text{ mg/m}^3)(\frac{1}{6}) = 0.29 \text{ mg/m}^3$   $(2) (1.6 \text{ m}^3/h)$ 

0.29 mg x 1.8 m3 x 2hr = 1.05 mg

day

day

TCDD @ 94 mg TCDD/Kg 2,4,5-T

1.05 mg x 94mg TCOD x Rg
106mg

 $= 9.87 \times 10^{-5} \text{ mg TCDD}$   $9.87 \times 10^{-2} \text{ ug}$ 

HD

average Case for HP

TCDD = 32 mg TCDD/Kg T

10% shin abe
100% canopy penet.

2.4.5-T Same

Dermal Inhal

9,3 mg 1,05 mg 10,35 mg

For 70 Kg man

0.148 mg 7

TCD D

Dormal

9.3 mg T x 32 mg TUDB X Kg T

= 0.29 Mg TCDD

Inhal

1.05 mg T x 32 mg TCOO x Kg 105 mg T x 105 mg

= 3.36 ×10-2 Ug

46

Inhal + Dermal Ego:

2.4.5-7

70

0.148 mg T

TODD

Donnal 0.29 Ug Mhal 0.03 0.32 Ug

70 =

0.004 llg TCAD

Worst Case for HO 14 mg TCOD/Kg T TCDD = 2,4,5-7 0.148 mg/ TODD 0.148/mg T 14 mg TUDIS Dermal 9.3 mg T x 14 mg TCAD x Kg = 0.13Mg TCDD

Inhal  $1.05 \times \frac{14}{106} =$   $1.47 \times 10^{-5} \text{ mg}$   $1.47 \times 10^{-2} \text{ Ug}$ 

MO Inhal + Dumal 2,4,5-7

0.148 mg T Kg

TCDD

Dumal

0.13 Ug 0.01 0.14 Ug TCDD Kg

\* Gas Chrometographie estimate DDT TODA ZX 10-7 1×10-70 Vapor Presence 0.2 ppb Water Sol. 1.2 ppb (~ 200,000) **2**40,000 Octand/water 1.4 × 106 1-6×106 Soil Movement Migration Tendency WATER - Solds Dicamba 1,3 - D EDB - 2, 4,5-7 Monum WATER - Air at aqualibrem 1,3-12 18 TCOD ( 115 DOT 130 Soil volatilité Are = water = soil TOD = 9×10-5

Pertinent Physical an Chemical A Chemical Factors alpeding Exposure to Herbicide Orange Water Insoluble formulation (specific growty = 1.28)
Concentrated Formulation (18-16 ai/91) (1978): 30 VAPOR RESSURE ~ 3.60 × 10 mm Hg
at droplet surface 0.01263 dynes/cm² Noncorroture to metal. Deleterous to paints, rubber and neopene \$5. Viscous (43 à 23°C) Long Shelve 1,40 There factors influence Performance in application agreement Denteraction of the spray with the faget of I final fate in the enveronment. Viscouty of aerosal 00185 antipose Serface tension = - Chloroform = 27.14 dynes/cm - 73 dynes/cm water = 73 dynes/cm Butarel = 25 dynes/cm

Sunlight 1.83ppb 18000141 21-0/silved (4 PM) 0.5 pb 0.041 DAY Z (10 AM) 0.26 (86% degraded (4PM) 7.5 ppm TCDDie Silver 2,800 pt Day 0 BE 951 Pay 24 Ty = 4 days

HATICLE SIZE : quality 500 76,2 21.9 Rate of Fall I min (n 350 el partice (2.5-ft/sec 30 min 2004 - 259 as very at 250 ft Typically - multicanopy forest 00 gallons dissemiated on 346 acres 940 Deposited on canopy at good level actual ground level deposition 0.17 gal/A or 18 lbs 34.0/2,457

NA 454-1 C-123 Accorde Delivery 130 KTAS 150 Ft altitude Droplet size 100 m to 500 m Sweeth 260 ft 1 em = 10 mm= 1000 p average drop = 135 aux = 175 m 0.01263 dymes = 2 (T) cm = 0.0175cm (0.01263)(00175) = T0.000 221

 $\frac{2}{2}$   $\frac{2}{1.1 \times 10^{-4}} = 7$ 

23=8 1 ft = 30.48 cm

average man = 70169 that. 6'= 182.9cm Height

Body Surface area =  $(\omega)^{0.425} \times (H)^{0.725} \times (0.07184)$   $(70)^{0.425} \times (183)^{0.725} \times (0.07184)$   $(6.08) \times 43.68 \times .07184$   $= 19.08 \text{ m}^2$   $1.5 - 2.0 \text{ m}^2 \text{ fm Fig 661}$ 

30°C= 60°F

VAPOR PRESSURE & at 20°C

NBE 2.4-D = 4 1.2×10<sup>4</sup>

NBE 2.45-T = 0.15×10<sup>-4</sup>

OTANGER - 30°C = 3.6×10<sup>-4</sup> mm Hg.

at 30°C ( D = 3.1×10<sup>-4</sup>

T = 0.45×10<sup>-4</sup>

20°C = 1×10<sup>-1</sup> TCDD 40 TEDD

T = 0 ange = 300 TEDD

350 particle size 2-3 Et sect 150 ft 1 min ]

- all fluids possess a definite resistance to change of form. This property, a sort of internal fruction is called viscosity; it so expressed in dyne-secondo/em² Teh land - un O. G at 20 C. water 1. Of Bestanol Bestanol Orange Lytt Madina ail 40 900 - 100°C = 129 Mp Butand = 140 = 125 Weiten Vapor Pressure - # 200 Heroury 8.001.2 (1.2x103) / Wate = 17,54 mm HZ Dresel Oil = 22 mm 3.6 × 10-4 mm fræsure exerted when at liquid is in equilibrium with its own vapor

350 M 7 mm 70 Kg mon 180'cm = ht 1.9m 2 = 1.9x10 min 0.660 HO × 1.9×106 mm 2 = = 1.63 ×10 5 UP HO on a mude ( 0.16 C HO 7-7 gms 2,4,7 .016 v 0.08 ET + 0,08 ED Calculate TEDD Cesing ETEDD] = 2 mg/kg 2 mg TCDD × 10.7# × 79 × 3.785 € × 2.57 Mg TCDD Worst Core - Mude 0,16 P HO X 2,57 M9 0.4/mg



tor 70 kg man - this represents done of 0.41 mg TCDD X 5.87 × 10-3 mg = 6.0 Ug Assume absorption @ 120 mg Kg. Day which is 120 X

Put clothing on hun such that exposed their is only 10% of Total es ou show is 0. 41 mg 7CDD x 0.1 0.041 mg 0.041 = 5.87×10-4 mg/kg 0.649/Kg assume 2% absorpt 600 Mg x .02 = (12 Mg)

Kmg 2,4.5-T 3.556 mg Mal. 0.46 # : Acre (12# Acre 2,4,5.T) (3.6)(12) ~ 93 mg 2.4.5.T 10% obser 9.3mg 2.4.5-1 Dore would be for TCDD is 94 mg TCDD

Kg T 0,13 mg T × 94 mg TCDD X Kg When X Kg T = 1,25 ×10-5 mg 1,25 XB-2 Ug 12.5 Mg

ronge & Deposit

was 1gol HD

from Vicherly's data

worst case (Ch 1)

D 2-31

Use 21% penutution

Using RPAR approach & 40 applie sate 10 gol 10 gal Oscure 40 # ac 2,4,57 Cessure. 12#T We applied

# Calculations of TCDD exposures

Assumptions:

Concentration of TCDD in HO Mean of 2.0 mg/kg

Concentration of TCDD in 2,4,5-T

Mean of 4.0 mg/Kg 2,4,5-T

According to Ramsey, Lavy & Braun worker exposure to 2,4,5-T as mixER in a 10 hour period 13:

O.13 mg 2.4,5-T (Kg B.W) (10 hours)

O.13 mg T × 4 Ng TCDD = ,052 Ng TCDD

Kg BW 10hrs mg T Kg BW · Hour

Hours TCDD Exposure (N3/Kq)
Thalken 1056 54.9
Tremblay 1046 54.4
Young 6608 343.6

Assuming 10% of B.W. 13 fat É 10% of Exposure dose 15 deposited

54.9 Ng TCDD x .1 BW x Deposited

Kg BW X BF x .1 Gxp

= 54.9 Ng TCDD / Kg B.F.

i.e. ~55 ppt Thalken
55 " Tremblay
344" Young

95% Conf. Cimits 
$$N = 0.046$$
 $M = 8$ 
 $V = N-1 = 7$ 

95% Upper Imit  $2 = 1.90$ 

90 "  $2 = 1.42$ 
 $\sqrt{X} = 2 = 1.42$ 
 $\sqrt{X} = 2 = 1.42$ 
 $\sqrt{Y} = 2 = 1.90$ 

95%  $= \sqrt{Y} + \frac{(1.9)(.046)}{\sqrt{7}}$ 
 $= \sqrt{2.645}$ 
 $\sqrt{Y} = 0.033$ 
 $= 0.106$ 

90%  $\sqrt{Y} + 1.42(.046)$ 
 $= \sqrt{7}$ 
 $= 2.645$ 

x + 0,024

(= .097)

at a conc of 4 Ng TCDD / mg 3,4,5-T  $0.2 \text{ mg } 2,4,5-T \times 4 \text{Ng TCDD} = 0.8 \text{ Ng TCDD} / \text{Kg BW · hour}$   $0.3 \text{ Kg BW · Hour} / \text{mg} = \frac{0.8 \text{ Ng TCDD}}{\text{Kg BW · hour}} / \text{Kg BW · hour}$  0.3 Ng TCDD / Kg Re / Kg BW · hourThat en 0.56 hr = 844 / Tremblay / 0.46 / 837 / 837 / 837 / 8386Assuming 0.608 / 5,286 / 5,286 / 6608 / 5,286 / 6608 / 5,286 / 6608 / 66

Assuming 10% of B.W. 15 fat

\$ 10% of exp. dose 15

Teposited

Thalken Tremblay Young 544 ppt 837 " 5,286 "

2,45-T Dose (mg/kg) 20017 Job Alliker Mox BOLKPOCK My Sporay 0.085 0.132 Tractor Mix. 0.053 0.086 0.054 0.041 Melic (Micro) Mix 0.081 0.092 Pilat 0.002 0.012 Helic (Rain) My 0.086 <u>a156</u> Pilat 0.049 0.046

### CAlculations of TCDD

× mg 2,4,5-T/kg body not/10 hr exposure 4 mg TCDD = 49, ng Kg 2,4,5-T = 8 mg Ransey, Larry E. Sraw. for mixER of herbicales 0.13 mg 2,4,5-7 x 4 mg 70D Eq BW.10 hes mg 2,4,5-7 O. 52 ng Tood Eg Bw. 10 hes Come Theires 1056he : Total dose = 10.56 x 52 mg = 549 mg TeDD assume 10% of BW = Boly for Ther, in body fat well have 5.49 rg/kg B.F. For Young 6,608 66.08 x 52 ng = 3436 ng/kg. Asserme 10% of BW = Boly Fat 30 mg/kg/ Them: 34 mg/kg BF BF Should Contain 34 PPT

Inhalation Dermal Efference : Total Shin, mg 2,4.5.T/Kg mg 2,4,5-T/Kg 0.03 Joseph / mx Treater 1/ mix 1.624 11.7 0.000191 Murfal /3 My 0.95 0.122 Roundrey / 8 My 0.40 0,042 13,25 Mean mg T 0.45 Exposure time was 2-3 hours .. Barel en Dermal exporure of

0.5 mg 2,4,5-T

Kg BW. 2.5 hours skind

> 0.2 mg 2,4,5-T (Kq.BW) (hour)

al, Marketines.

**F**2

Cereme that in walking them an area just heated that a man would have his feet, Ecafo exposed to 40. (24% of total body whin surface) The level would be based on his reaching equilibrium w amt deposited/m²

18 Capplication rate = 3 gol HD/Acre

Penetration to ground = 10% = 0,30 gol/ane

0+8 gol +3.7850

2.4.5-T = 4 #/gol = 1,2 #/acre

0.55 Kg , acre = 1,35 X10 - 4 Kg

acre 4,047 m² m²

1.35 ×10-19/m2 = 0.1359/m2

Oliverge man har 1.9 m²
Shin area affected = 0.46 m²
24% of 1.9 m² fected = 0.46 m²

: The 2.4.5-Ton whine would be

0.135g T x 0.46m² = 62 mg T m2 x man

Grunne 10% absorption 19/Kg (0.1)(62) = 6.2 mgT

0.18 TCDD @ 2 mg/kg ug/g mg/mg 12.4 mg/mm

0.89 @ 10 62 "

1.77 @ 20 124 "

8.3 0 94 582 1

With a mean droplet diameter of 350 st one ran determine from that that a total of O.6 sel of HO would cause an area of approximately 70 mm? For an average man of 70 kg, height of 180 cm the total whin surface area would be 1.9 × 10° mm? (Du Boir)

Assuming 21% penetrates cover:

Assuming 10% absorbed

(1.615)(.1) = 0.1629 2,4,5-T

Man

abunda 0.162g 2,4,5-Tx 94mg TCDD x Kg x man man Kg T 1000g 70Kg

 $= 2.18 \times 10^{-4} \text{mg} = 2.18 \times 10^{-1} \text{ug}$  = 0.22 ug  $= \frac{0.22 \text{ug}}{\text{kg} \cdot \text{Day}}$ 

ZZO Mg Kg. Day 0.008e 3.785e

A-Z

1) 
$$\frac{0.6 \text{ ul}}{7.0 \text{ mm}^2} \times \frac{1.9 \times 10^6 \text{ mm}^2}{\text{man}} = 1.63 \times 10^5 \text{ ul HO}$$
  
= 1.63 \times 10^2 \text{ml HO}  
= 0.016 \text{ l HO}

2) with a TCDD worst case of 47 mg TCDD/Kg HO

= 60,39 mg TCDD/e HD

3) TCDD dermal exposure

4) 2,4.5-T dermal exposure:

$$0.008R = 2.11 \times 10^{-3} \text{ gal}$$
  
 $2.11 \times 10^{-3} \text{ gal} \times 4 + \frac{40R}{9aR} + \frac{2.45}{9aR} = 5.454 + 2.457$ 

 $\begin{cases} 8.454 \times 10^{-3} \# 2,4,5-T \times \frac{Kg}{2,2\#} \\ = 3.84 \times 10^{-3} \text{ Kg} = 3.84 \end{cases}$ = 3,84g 2,4,5-T 0.016 R HO x gal x 4# 2,45.T x 29
3.785 e 4# 2,45.T x 29 = 7.686 ×10 ×9 2.4.5-T Checking TCDD exposure 7,089 2,4.5-T x 94mg TCDD x Kg 0.721 mg TCDD

(head, shoulders, neck, foregrowe, hard Ethighs) are is affected,
7.689 2.4.5-T (0.1) Shin

man (0,1) or

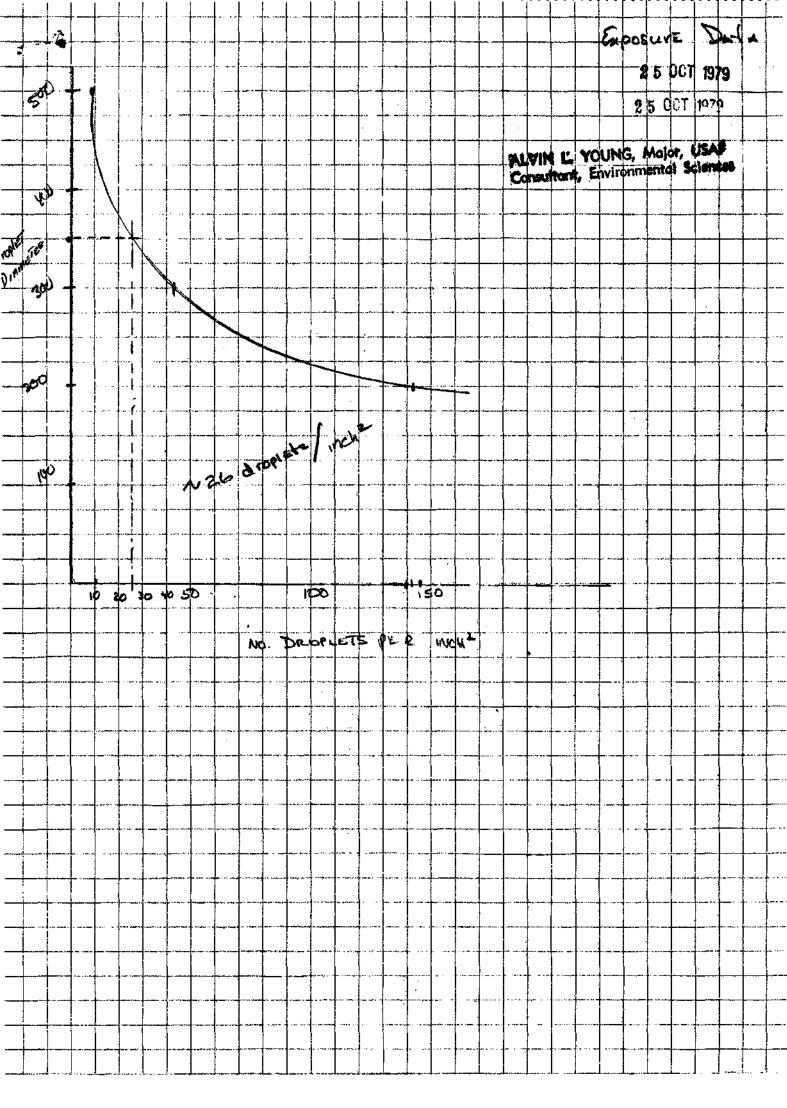


TABLE 1. Effect of Atomization on Coverage (a). Rate of Application i gallon per acre.

Droplet Diameter (microns)	<ul> <li>No. spray droplets per square inch</li> </ul>
10	1,148,000
20	143,000
50	9,224
100	1,164
200	142
300	43
500	9

(a) from Himel (1969) p. 920

sect control was the effect of low relative humidity. When the humidity dropped below 70%, insect control declined. In 1977, a comparatively humid year, excellent insect control was obtained with all of the standard insecticides. It was determined that rapid evaporation of the smaller, insecticidally effective droplets was a major factor in poor insect control.

Droplet size is of major importance in insect control. Very fine droplets (less than 50 microns or 1/500th inch in diameter) of short residual insecticides appear to be highly effective in controlling insects. These small droplets impinge on the setae and other parts of the insect's body. A larger droplet may strike the pest causing mortality; however, the smaller droplets increase the probability of hitting the target pest. According to Himel and Moore<sup>1</sup>, 93 percent of the mortality of tobacco budworm (Heliothis virescens), cabbage looper (Trichoplusia ni) and boll weevil (Anthonomus grandis) was caused by droplets less than 50 microns in diameter. Later, Himel2 reported that if the optimum droplet size is around 20 microns, then our present spray efficiency is about 1% or less.

Of almost equal importance is the effect of droplet size on coverage and penetration of the canopy. Higher volumes of spray are generally recommended in an effort to obtain more complete coverage and better penetration of the plant canopy. It is true that increasing volume will increase numbers of droplets and, therefore, coverage. However, since coverage is primarily a function of numbers of

droplets, a more practical way to increase it is to break the spray up into small drops. Each time the average drop size is divided in half, the number of drops is increased approximately 8 times. These data are presented in Table 1. Thus, one gallon of spray per acre divided into 50 micron drops will produce the same coverage as 64 gallons applied in 200 micron drops.

A third major advantage of very fine droplet sprays is their ability to penetrate through very small openings in the bud tissues and beneath bracts enclosing squares and bolls where the bollworm and boll weevil live and feed. Additionally, fine droplets, because of their tendency to float on air currents, penetrate plant canopies more efficiently, are deposited on the undersides of the leaf, and deliver more insecticide to the lower portions of the plant. This gets the insecticide out of the sunlight and slows the rate of ultraviolet degradation. This in turn, extends the efficacy and increases the control of the bollworm and budworm moths that spend the day down in the canopy.

Since mites and whiteflies live almost exclusively on the undersides of the leaf, only fine droplet sprays can reach them.

Larger drops, due to their greater verticle velocity, tend to impinge on the upper surface of the leaves.

Despite the many important advantages of small particle sprays, there are a number of serious disadvantages. The most serious of these is the rapid rate of evaporation of the insecticide carriers. In practice, this is almost

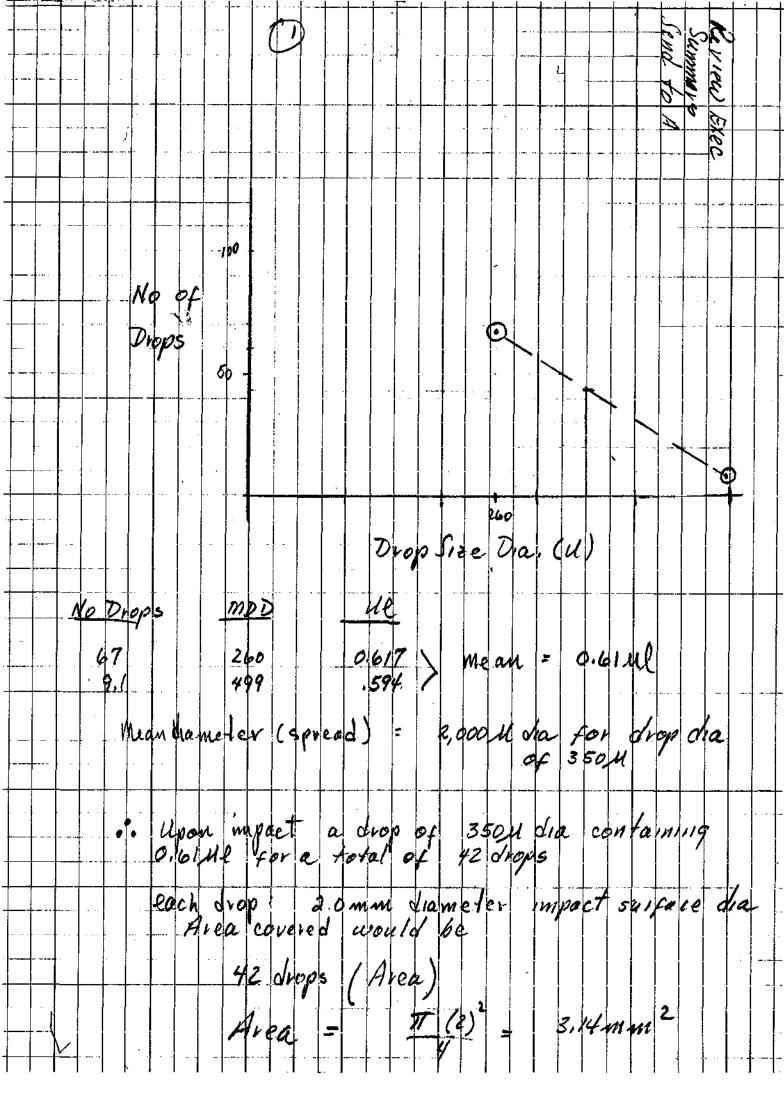
universally water. Seymour and Byrd<sup>3</sup> projected that an 80 micron water droplet falling through air at 70% relative humidity will disappear in approximately 8.5 seconds. During that time, the 80 micron drop will fall less than 2 feet! A 50 micron drop will last only about 2 seconds. Figure 1 shows this relationship. Boise4 reports that evaporation and figuid density are often overlooked as factors contributing to the drift of sprays. A spray mixture that is 95% water will evaporate readily, with a 100 micron diameter drop being reduced to a 40 micron drop in about 15 seconds when the air temperature is 80°F and relative humidity is 50%.

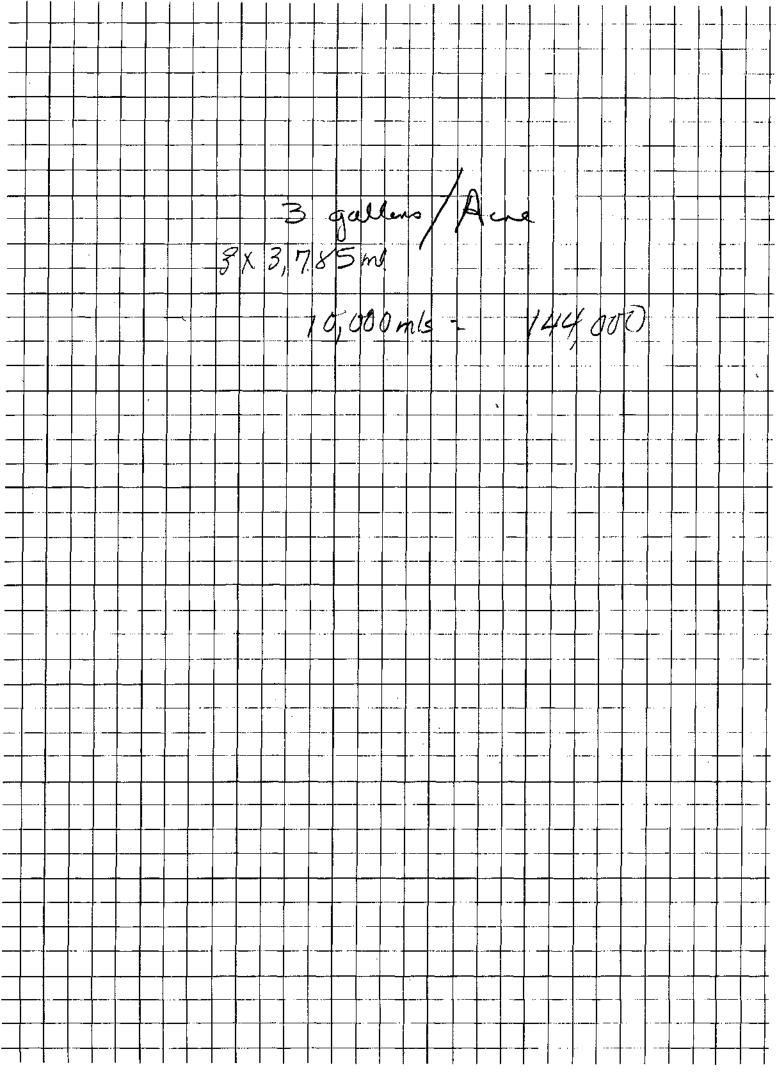
Even though the solvents and emulsifiers in the insecticidal formulations reduce the rate of evaporation somewhat, it is still extremely difficult to get an appreciable percentage of the applied insecticide into a cotton crop in an effective droplet size range as long as water is used as the principal carrier. Thus, evaporation can be reduced by using low-volatile spray carriers in lieu of water.

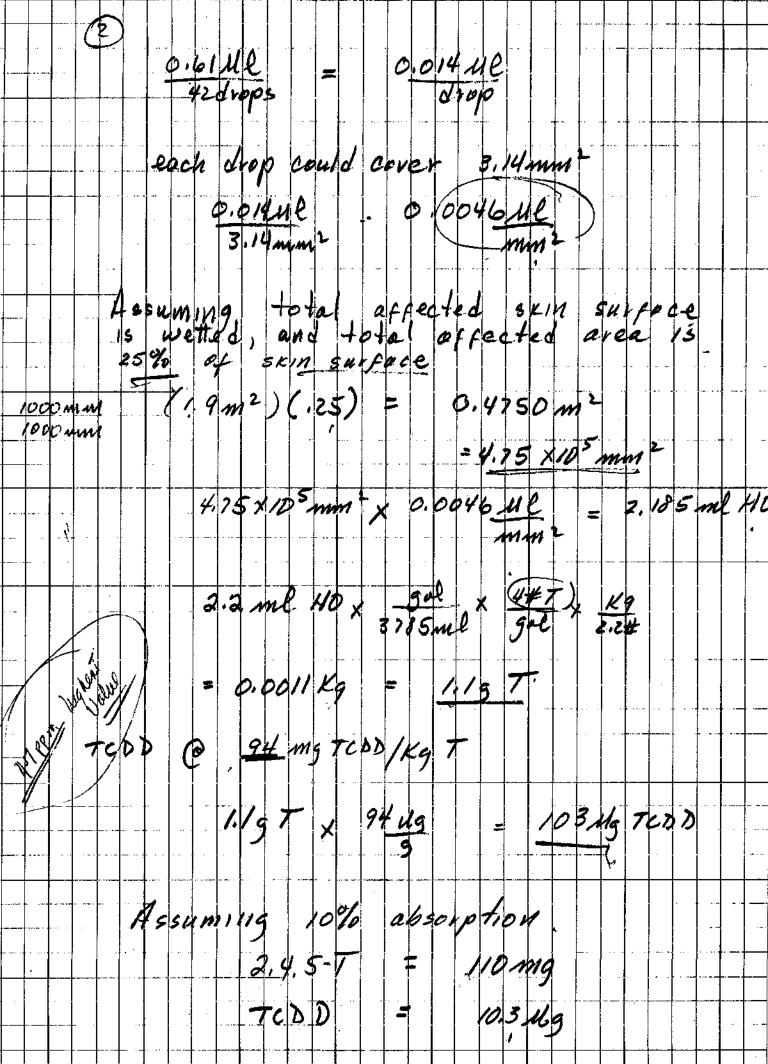
A large number of low volatile materials may be used as substitutes for water as insecticide carriers. Wrights suggested that when concerned with evaporation on a real hot, dry day, to add some propylene glycol or even ethylene glycol to the spray mixture. In the initial screening a number of criteria were used for selecting candidate carriers for inclusion in these studies:

- 1, relatively non-volatile
- 2. non-phytotoxic
- 3. compatible with insecticides
- 4. non-corrosive and non-damaging to aircraft components
- non-flammable
- 6. relatively non-toxic
- 7. physical characteristics as close to water as possible or at least sprayable through systems in current use without major modifications
- inexpensive (or their cost to be more than offset through reduction of insecticide required)
- 9, cleared for use on crops

Small plot studies on two month old cotton were conducted during 1978 on insecticidal formulations containing







70 Kg Man a 1.57mg/kg 2.4.5-748 D 147 19/19 4 mg tcos/kg 7622 4.4.497600 1.19 X 10% absorpt 0.44213 74100 0063 m/kg 0.44Mg 70 kg man EPA -- AD -- AN -- AVINALO

HEAN FOR Orange 367 m.m.

Hust & Damou DROP mmD ×126 260 498 # trops in = 400m-500m = 100 drops/in2 1.5 SA METER = 2325 In2 381 po grange = 361 min 88.582500 grams incepted body 4 191 mg x.088 to = .17 mg Tedal
4 90 kg man = .17 mg = .0024 mg/kg.

### Textbook of

## MEDICAL PHYSIOLOGY

SECOND EDITION, Illustrated

#### ARTHUR C. GUYTON, M.D.

Professor and Chairman of the Department of Physiology and Biophysics, University of Mississippi School of Medicine

1961

W. B. SAUNDERS COMPANY

Philadelphia and London

is not the reason why the basal metabolic rates of different persons vary according to the surface area; instead, this relationship is only an empirical one.

Referring once again to Figure 660 it will be noted that the total number of Calories liberated by the patient per hour is divided by his total body surface area of 1.5 square meters. This means that his basal metabolic rate is 48.3 Calories per square meter per hour.

Method for calculating the total surface area. The surface area of the body varies approximately in proportion to weight <sup>0.67</sup>. However, more accurate measurements of the body surface area have shown that it can be determined more accurately by a complicated formula based on weight and height of the subject as follows:

Body surface area =

Weight 0.425 × Height 0.725 × 0.07184 Figure 661 presents a graph based on this formula. In the formula and in the figure, body surface area is expressed in square meters, weight in kilograms, and height in centimeters.

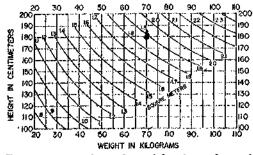


Figure 661. Relationship of height and weight to body surface area. (From DuBois: Metabolism in Health and Disease. Lea & Febiger.)

Expression of Basal Metabolic Rate in Torms of Weight. Measurement of the basal metabolic rates of many different species of animals has shown that the rates do not vary precisely in proportion to the body surface area. Instead, in animals ranging in size from the mouse to the horse, the basal metabolic rate has been found to be proportional to weight 0.784. Because surface area is approximately proportional to weight 0.67, it is obvious that correlating basal metabolic rates between animal species on the basis of surface areas would be in extreme error. This fact has considerable implication in human physiology and in clinical medicine, for some physiologists believe that even in comparing basal metabolic rates between human beings the factor weight 0.784 should be used instead of surface area. If this is true, overweight subjects would have to have basal metabolic rates considerably above the mean as based on the surface area method in order to be normal, and very thin subjects would have to have basal metabolic rates considerably less than the mean as based on the surface area method in order to be normal.

Expression of Basal Metabolic Rate in Percentage Above or Below Normal. In Fig. 659 it will be noted that the basal metabolic rate varies tremendously with age; also, males in general have a basal metabolic rate approxi mately 8 per cent greater per square meter than that of females. Therefore, to compare the basal metabolic rate of any one subject with the normal basal metabolic rate, it is necessary to refer to a chart such as that in Figure 659, which gives the normal basal metabolic rate per square meter at each age and for each sex. Once reference has been made to such a chart, the basal metabolic rate is ordinarily expressed as a percentage above or below normal. For example, in Figure 659 the normal basal metabolic rate for a 20-year-old male is shown to be 38.5 Calories per square meter per hour. Therefore, if the particular patient represented in the calculations of Figure 660 is a 20-year-old male, he liberates 9.8 Calories per square meter per hour above the normal mean value. It is then determined that this is 25.5 per cent above normal. Therefore, the basal metabolic rate is expressed as plus 25.5. Similarly. basal metabolic rates below normal are expressed as minus values.

Constancy of Basal Metabolic Rate in the Same Person. Basal metabolic rates have been measured in many subjects at repeated intervals for as long as 20 or more years. As long as a subject remains healthy, almost invariably his basal metabolic rate as expressed in percentage of normal does not vary more than 5 to 10 per cent.

Constancy of Basal Metabolic Rate from Person to Person. When the basal metabolic rate is measured in a wide variety of different persons and comparisons are made within single age, weight, and sex groups, 85 per cent of normal persons have been found to have basal metabolic rates within 10 per cent of the mean. Thus, it is obvious that measurements of metabolic rates performed under basal conditions offer an excellent means for comparing the rates of metabolism from one person to another.

#### REFERENCES

Bozler, E.: Plasticity of contractile elements of muscle as studied in extracted muscle fibers. Am. J. Physiol., 171:359, 1952. Bozler, E.: The role of phosphocreatine and adenosinetriphosphate in muscular contraction. J. Gen. Physiol., 37:63, 1953.

Bozler, E., and Prince, J. T.: The control of energy release in extracted muscle fibers. J. Gen. Physiol.,

37:53, 1953,

Buchtal, F., Svensmark, O., and Rosenfalck, P.: Mechanical and chemical events in muscle contraction. Physiol. Rev., 36:503, 1956.

Chance, B.: Enzymes in action in living cells: the steady state of reduced pyridine nucleotides. Har-

vey Lect., 49:145, 1953-1954.

Conway, E. L.: Nature and significance of concentration relations of potassium and sodium ions in skeletal musele. *Physiol. Rev.*, 37:84, 1957.

Crowell, J. W.: A continuous recording oxygen debt

analyzer, Fed. Proc., 19:102, 1960.

Drummond, G. I., and Black, E. C.: Comparative physiology: fuel of muscle metabolism. Ann. Rev. of Physiol., 22:169, 1960.

Greenberg, D. M.: Chemical Pathways of Metabolism. New York, Academic Press, 1954.

Guyton, A. C., and Farish, C. A.: A rapidly responding continuous oxygen consumption recorder. J. App. Physiol., 14:143, 1959.

Hasselbach, W., and Weber, A.: Models for the study of the contraction of muscle and of cell protoplasm. Pharmacol. Rev., 7:97, 1955.

Hsia, D. Y.; Inborn Errors of Metabolism, Chicago, Year Book Publishers, 1959.

Huxley, A. F.: Local activation of striated muscle from the frog and the crab. J. Physiol., 135:171, 1957.

Huxley, A. F.: Muscle structure and theories of contraction. Prog. Blophys., 7:255, 1957.

Huxley, A. F., and Niedergerke, R.: Measurement of muscle strictions in stretch and contraction. J. Physiol., 124:461, 1954.

Huxley, A. F., and Niedergerke, R.: Structural changes in muscle during contraction; interference microscopy of living muscle fibers. Nature, 173: 971, 1954.

Huxley, A. F., and Taylor, R. E.: Function of Krause's membrane. Nature, 176:1068, 1955.

Klotz, I. M.: Some Principles of Energeties in Biochemical Reactions, New York, Academic Press,

Laidler, K. J.: Introduction to the Chemistry of En-

zymes. New York, McGraw-Hill Book Co., Inc.,

Lipmann, F.: Biosynthetic mechanisms. Harvey Lect., *44*:99, 1948–1949.

Mommaerts, W. F.: Investigation of the presumed breakdown of adenosine-triphosphate and phosphocreatine during a single muscle twitch. Am. J. Physiol., 182:585, 1955.

Mommaerts, W. F.: Is adenosine triphosphate broken down during a single muscle twitch? Nature, 174:

1083, 1954,

Mommaerts, W. F.: The effect of adenosine triphosphate upon actomyosin solutions, studied with a recording dual beam light-seattering photometer. J. Gen. Physiol., 39:821, 1956.

Mommaerts, W. F.; The proteins of muscle and their participation in the process of contraction. Am. J.

Phys. Med., 34:11,1955.

Mommaerts, W. F., and Hanson, J.: The effect upon actomyosin of stoichiometric amounts of adenosinetriphosphate regenerated in a coupled enzyme system. J. Gen. Physiol., 39:831, 1956.

Neilands, J. B., and Stumpf, P. K.: Outlines of Enzyme Chemistry. New York, John Wiley & Sons,

1955.

Pearl, D. C., Jr., Carlson, L. D., and Sherwood, W. W.: Mechanism of oxygen deficit. Proc. Soc. Exp. Biol. & Med., 92:277, 1956.

Perry, S. V.: Relation between chemical and contractile function and structure of skeletal muscle cell. Physiol. Rev., 36:1, 1956.

Szent-Gyorgyi, A. G.: Bioenergetics, New York, Academic Press, 1957.

Szent-Gyorgyi, A. G.: Structural and functional aspeets of myosin. Adv. Enzymol., 16:313, 1955.

Szent-Gyorgyi, A. G., Mazla, D., and Szent-Gyorgyi, A.: On the nature of the cross-striation of body muscle. Biochim. Biophys. Acta, 16:339, 1955.

Theorell, H., and Duve, C. Dé: Myohemoglobin. Arch. Biochem., 12:113, 1947.

Umbreit, W. W.: Metabolic Maps. Minneapolis, Burgess Publishing Co., 1952.

Watanabe, S., and Sleator, W., Jr.: EDTA relaxation of glycerol-treated muscle fibers, and the effects of magnesium, calcium and manganese ions. Arch. Biochem., 68:81, 1957.

Weber, H. H.: Adenosine triphosphate and motility of living systems. Harvey Lect., 49:37, 1953-1954.

HURTI Report Drop diameter -lotal NO-06 had ana Drops contacted ne 00 % 64 0.6 498 M 0.6 7010 0.75 127 2 260 4 0.62 67 0.59 (100 gal/sta) Average Tow Pale Average Swall Deag here Orcuse 227 GPH 259 364

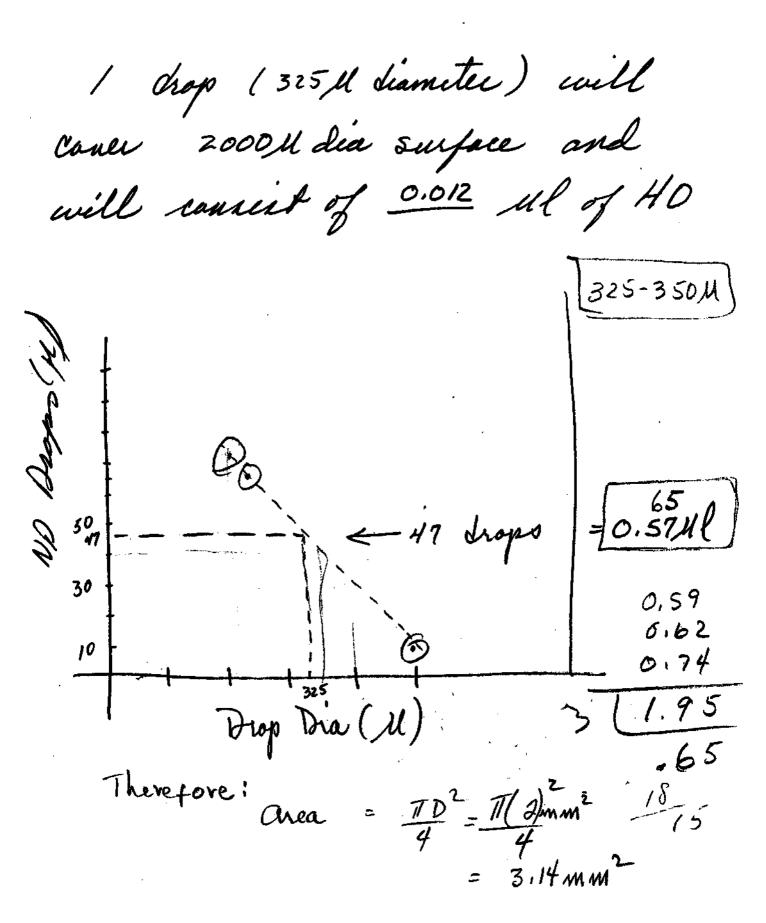
Aerial appl: RPAR—

Derinal— 5×10-6 Ug/Kg 7/9/0

Milol — 2×10-6 Ug/Kg 299/0

.

.



The body surface acea Afected is 31% of total area (47m²)(31) = 0.59m² - 5.9 X/0 mm2 for each 3.14 mm² whene 0.012 ul 5.9×10 mm × 0.012Ml = 2,082 Ml = 2.1 mlarruning 10% obsailion Total = 0,21ml 0,21 ml HOx gal x 4# x K9 x 3785 ml x gal x 2,2# x 1410-18 = 1.00 ×10-4 xq 2.4.5-T absorbed 19,00 mg 101#mg 2.4.5-T 1410-18

1.43 mg/Kg 245T 101 mg = 49/9 1.43 mg 2.4,5-T x 上對如 Mg/Kg.mam abrocked TCDD 2.86 2 15 21,3 42.6

90

10°U = 1 cm  $10^3M =$ 100 cm = 1 m 1,000 mm = 1m Drap dia
1,000,000 um = 1m Drap dia
Card Spot dia 1 M = drop dia x # drops Card spot dia = 495U Area =  $\frac{\pi D^2}{4}$ 6.006 X 9 Draps 540,55/ um 103 mm 1 mm² = 106 km²
5.4 ×10 cm² 1,750 M - 2,100 M Say 2,000 ll for each droplet of Size 325 K

tiskt!  $\mathbb{C}^{2}X$ ,)<u>,</u>

0.57Ml 125M 564 drops 0.57Ml 250M 70 drops " 500M 8.7"

area No Total Contacted ue 16.68 ,572 18.09 , 583 584 243 9.10 0.534 70.8 ,590 8.59 64.1 495 5.11 9.0 ,572

Spread factor = Drop dia ul Card gest chia

Card spot = 495 = 276 M

 $\frac{mm}{mm}$  area =  $\frac{77D^2}{4}$  -  $\frac{5.4055}{mm^2}$ 

M. M.

PERSONNE L

## COMPARISON OF RANCH HAND, AND GROUND TROOP EXPOSURE TO AGENT ORANGE

RANCH HAND

Console Operator:

VAPOR: PACER HO data (Air Force Technical Report 18-92, 1978) - within Dedrum Facility at Naval Construction Battation Center, Gulffort 18, 90-950 F.

- Are Concediation

 $2.4-D = 135 \text{ mg/m}^3$  $2.4.5-T = 80 \text{ mg/m}^3$ 

- Respiration rate for quenage man at works

The your CHANCE COTER OF

affork form 2.41.8 Hilliant (2.42.2) maintenance full (2.42.2)

Water/Air Ratio:

Water/Air = (Water 501) (8.206) (0.76) (273.15 + T°C)
(Vapor Press.) (Mol. wt.) (102)

water/Air = ppm/ug/cc

. Water oil. = ppm

Vapor Press. = mm Hgo

- No AOWG TR TO DOD 200 names from Unito Christia - 4400 Names a No Conscien Afort to exchade nonethin Service Exclude Officers EXCLUDE KIA TOURS · Engineer bestlaken - Constener

Nice Conventiony - Infansay Division 15,000.

Brights

3 Man. MARINES Delete Charolie Company Signal Lugare Medial Brigade Universe / Untram Vellagno

\* Infaling. - MOS Platom. Platon - 37 4 Date Inducation Rotates out of Vietnam. Entry on Seath Solodiand Subjects for the pilot phase. Reference Abstracts

Population, description all soldiers in qualiful Battali-Not Survey of the Population e.g. Two groups of high suposure RANCH HAND Chemical Corps # Cortera for enclusion & Exclusion © Cover major 1.14 of persons € - Draftees with one town of duty - One ye town Vietnam Statistical Problem - read to
over-sample
Strat fication Require

- Epi-study - compare health outcomes
situation two major groups. = Exclude Multiple towns. Officers may have more confoundars N= enlested groups

· Hust be in unit for 10 months.

· Hust track more carefully unexposed personnal. @ Wounded in Oction & Bares excus rist · Charactergation of Population. Mortalidy Facial
Racial - Education - Regional Salar Mice us 105 Mice Menters Rural/Urban

## Exposure Subcommittee

· Carl Keller - Hailing list for UA Health Effects Com

24-47 48-72

Minusers for Solybury He to Ayres 10 Ball HR to AROTT Australian Response Slides for Salsburg Kay (aroll Un to Ed Gangslead

SALZBARG -Denoti - Town Stely ( Hr. to DOD) copy to: House o Stand Chort Selection ) HIHS PETE Flynn. Her on 1,000 ? ) our Pilol place Subjectes of Recruitmen Phase I Rebuich Oversile te. Supe

Alorsony Commelles Epidenies

Chemical, Target, Time afte application, and distance from application site. Vapor Pressure o Disdemination of Herbicides
Application potameters
Target vegetestra o Route of Exposeno Inhalation Ingestion · Enveronmedal Fate en vegetation in water Dussemenation Systems. 1. RANCH HAND NA 45 V-1 Internal Defolion 2. UH - JB/D Helicopter Spray System ( Ag RINAUTICS) 3. Buffalo Turbine

4 Parer Driver Decontamination Apparation (PDDA)
5. Back - Pack Sprayers (245 gallon)
6. "Terry-regel" or Field Expedied Devices

A Marian

ASSIGNMENT OF WEIGHTS

Score WT Toxicity of Hebruide Orang Goomg/kg 2600 mg/kg White Blue 3100 mg/kg o Location of Musion De O wation Tungle (heavy (anopy) 10% paretration Mangrove (hight Campy) <del>3</del>0% 1808 ... Crop/Abox1 10 Verimete. 40% " Time / DISTANCE HATRIX Distance (km) TIME SPACE

A STATE OF THE STA

[030] EPA Set 28,29 RTP RH admining JUBCOMMITTEE Time & distance parameters · Weighting Scheme for exposure index @ Ryllman 2 Lose response US threshold Window Review Building Obtain appropriate Station LIST (My who Detains bathlier Station LIST (My who Detarnine availability of records Vecard unit doily from coordinates (activity) Nevelop In service 4 HERES TAPE Inlex of weighter,

days 3 2 - -

Jamp 70 15 10 5

18 1 3 1

Fate of TCDD Mission Location of Jungle Mangrove Type & Mission Involved RANCH HAND Perimeter Abore Turne after Exposure Days · within ce for Exposure Type (application Sito) 0.5 km 1.5 20

...

·--

See Stevenson Exposure Chort č ō Combat Orange Von Combat Craze 20 Non Combat Non Orange C 0 EXPESORE 1. Threshold VS GRADIETT Duano Z. INDEX Calculation (So cité /Political) 20 Hote of Orange 8-12 Intermediate
10-20 High Orange Location Weighting Scheme: Parmete by and distance

descripe "Non-HTT"
Destance US Timo